

PAR-1603010602020600 Seat No. _____

M. Sc. (Physics) (Sem. II) (CBCS) (W.E.F. 2016) Examination

August / September - 2020

CT-06: Atomic and Molecular Physics

Tim	e : 2	$\frac{1}{2}$ Hours] [Total Marks : 7	'0
Inst	truct	 ions: (1) All questions carry equal marks. (2) Full marks are indicated at the right side of eac question. (3) Symbols have their usual meanings. 	:h
1	Ansv	wer any seven of the following:	.4
	(a)	For the term symbol ${}^{2}P_{3/2}$ find the value of S, L and J.	
	(b)	Mention the two different methods by which one can sum orbital and spin momentum of several electrons.	
	(c)	How will you categorize Zeeman Effect?	
	(d)	State the interval rule.	
	(e) (f)	What are the rigid and the non-rigid rotators? In case of a diatomic molecule what is the relation among D, B and \overline{v} ?	
	(g)	What quantum numbers J and K represent?	
	(h)	What is zero-point energy?	
	(i)	Explain the subdivision of IR region showing the importance of each subdivision.	
	(j)	What is the role of "Monochromator" in IR Spectrophotometer?	
2	Ansv	wer any two of the following:	
	(a)	Discuss the fine structure of H-atom spectrum.	7
	(b)	Explain in detail how electrons occupy orbitals in	7
		large atoms.	
	(c)	Discuss the concept of (i) quantization of direction and	7
		(ii) spinning electron.	
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3	(A)	Discuss the procedure of summation of	7
		(i) orbital contributions and (ii) spin contributions.	
	(B)	Discuss the explanation of normal Zeeman Effect on	7
		the basis of vector atom model.	
		OR	
3	(A)	Discuss the interaction of radiation with rotating	7
		molecule.	
	(B)	Explain the isotope effect in rotational spectra. Also	7
		state that transition originating from the level having	
		which J value will have the maximum intensity.	
4	Ans	wer any two of the following:	
	(a)	What are prolate and oblate symmetric top molecules?	7
		Explain energy levels and transitions for the rigid	
		prolate and the rigid oblate symmetric rotors.	
	(b)	Explain the working of a Microwave Spectrometer	7
		in detail.	
	(c)	Discuss the Morse curve and the energy levels of a	7
		diatomic molecule.	
5	Write short-notes on any two of the following:		
	(a)	Vibrating Diatomic Molecule	7
	(b)	IR Spectrophotometer	7
	(c)	Normal Vibrations of CO_2 and H_2O molecules	7
	(d)	Paschen Back Effect in the principal series of Lithium	7